

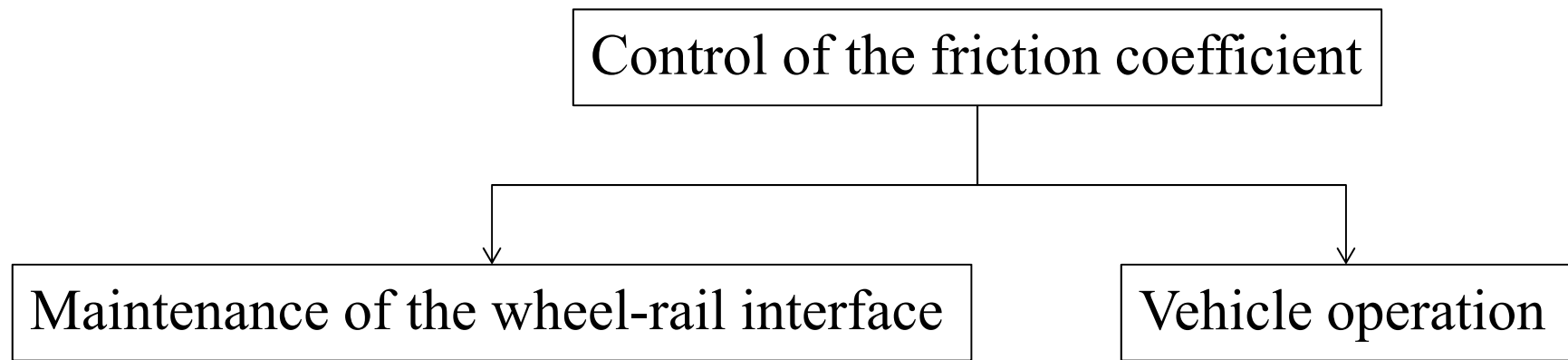
A Pin-on-disc Study of the Rusts on Wheel Rail Material and its influence on the friction coefficient



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- Importance
- Influential factors: environment; third-body; material properties...
- Measures: friction modifier; lubricant; grinding

Rust and oxide layers



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What is rust?

Rust is hydrated iron oxides.

Iron oxides refer to anhydrous iron oxides.



TABLE 1—PROPERTIES OF IRON OXIDES AND RUSTS

TYPE	MINERAL NAME	CHEMICAL NAME	FORMULA	COLOR
Anhydrous	Wustite	Ferrous oxide	FeO	Black
	Magnetite	Ferrosoferric	Fe ₃ O ₄	Black
	Martite	Ferric oxide	Fe ₂ O ₃	Black
	Hematite	Ferric oxide	alpha-Fe ₂ O ₃	Orange Red
	Maghemite	Ferric oxide	gamma-Fe ₂ O ₃	Reddish Brown
Hydrated (RUSTS)	Goethite	Ferric hydroxide	alpha-Fe ₂ O ₃ .H ₂ O	Red-brown
	Akaganeite	Ferric hydroxide	beta-Fe ₂ O ₃ .H ₂ O	Red-brown
	Lepidocrocite	Ferric hydroxide	gamma-Fe ₂ O ₃ .H ₂ O	Yellowish-Red-brown
	Feroxyhyte	Ferric hydroxide	delta-Fe ₂ O ₃ .H ₂ O	Red-brown
	Xanthosiderite	Ferric hydroxide	Fe ₂ O ₃ .2H ₂ O	Red-brown
	Esmeraldite	Ferric hydroxide	Fe ₂ O ₃ .4H ₂ O	Red-brown
	Turgite	Ferric hydroxide	2Fe ₂ O ₃ .H ₂ O	Red-brown
	Limonite	Ferric hydroxide	2Fe ₂ O ₃ .3H ₂ O	Yellow to reddish
	Hydrogoethite	Ferric hydroxide	3Fe ₂ O ₃ .4H ₂ O	Red-brown
	Ferrihydrite	Ferric hydroxide	5Fe ₂ O ₃ .9H ₂ O	Yellow-brown



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----D. Godfrey, "Iron oxides and rust in tribology", Technical paper, Journal of the society of tribologists and lubrication engineers 1999; 55(2); 33-37

Rusts on rails:

Different rusts and iron oxides on the rail can result in various friction coefficient causing stick-slip which leads to rail corrugations.

---M. Ishida, et. al, "Rail corrugations caused by low coefficient of friction in a submarine railway tunnel".



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It was reported that β - $\text{Fe}_2\text{O}_3 \cdot \text{H}_2\text{O}$ which is generated under the moist environment with the presence of chloride (Cl) has a great impact on reducing friction coefficient. A detailed surface treatment was also suggested.

---J. Suzumura, et. al, "In situ X-ray analytical study on the alteration process of iron oxide layers at the railhead surface while under railway traffic", Wear 271 (2011) 47-53

Two exposures:(pins and discs are made of wheel rail materials)

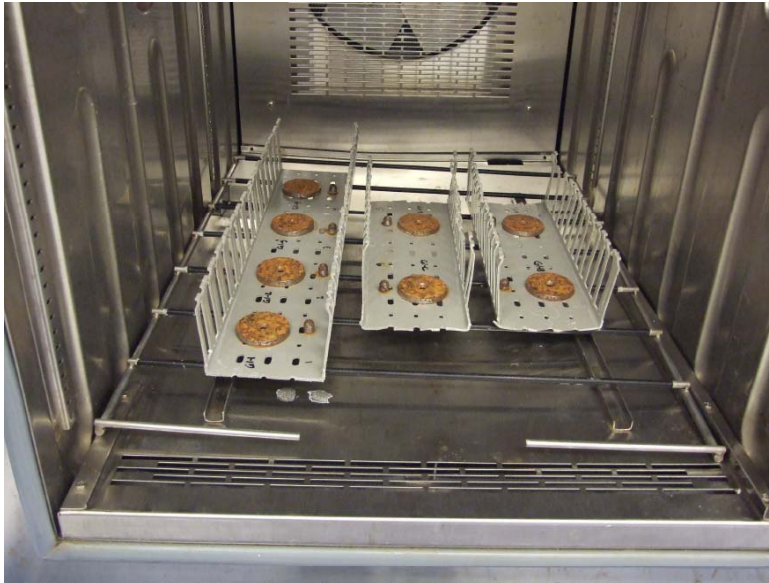


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	Group1 G1 (five cycles)	Group 2 for comparison G2 (five cycles)
Step 1	exposure in air for 0.5 h at 40 °C and RH> 90%	exposure in air for 2 hours at 40 °C and RH > 90%
Step 2	exposure in air for 0.5 h with 1% NaCl solution spray at 40 °C and RH> 90%	exposure in air for 6 hours at 40 °C and RH< 30%
Step 3	exposure in air for 1 h at 40 °C and RH> 90%	
Step 4	exposure in air for 6 h at 40 °C and RH< 30%	



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G1 exposure



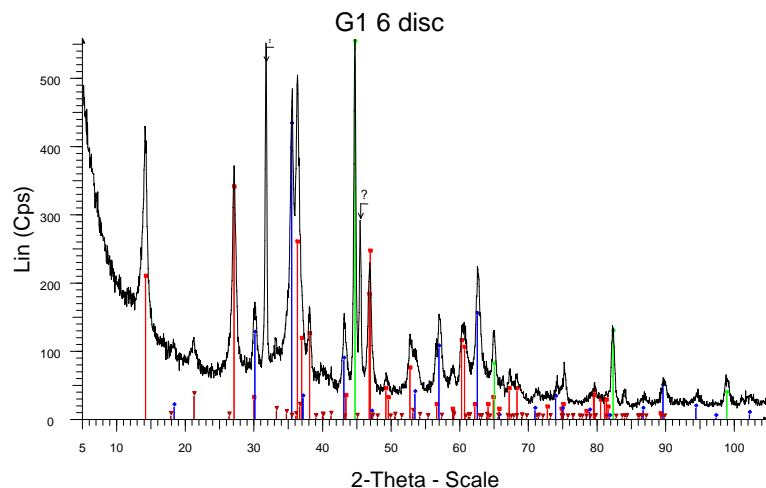
G2 exposure



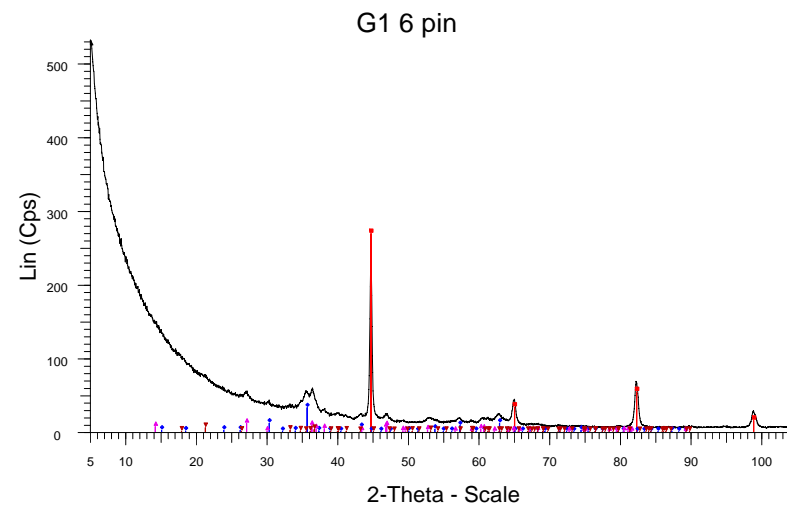
XRD X-ray diffraction analysis



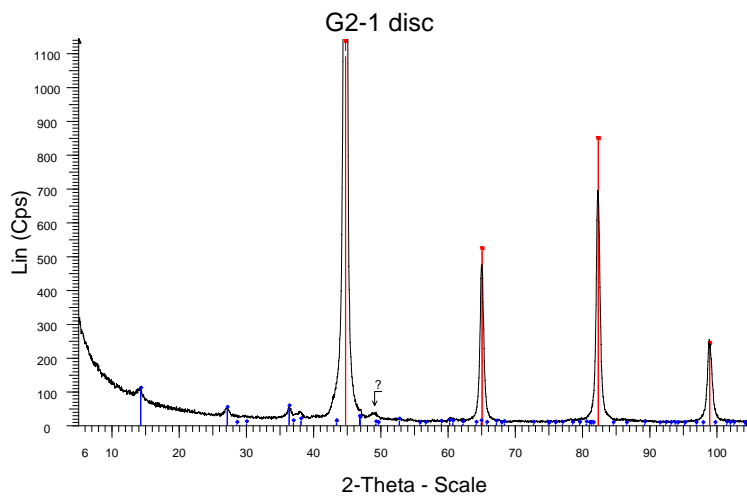
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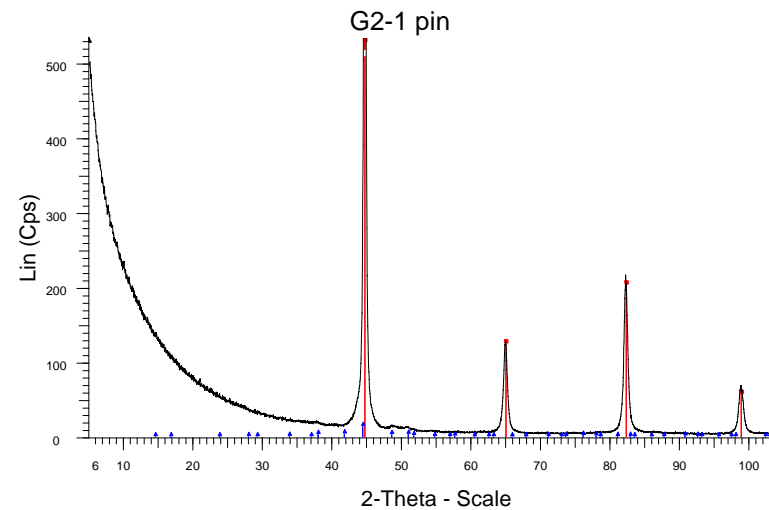
- G1 6 disc
- 00-044-1415 (*) - Lepidocrocite, syn - Fe+3O(OH) - Orthorhombic - Base-centered - Bbmm (63) - I/c PDF 1.2 - S-Q 65
 - 04-009-2284 (*) - Magnetite, syn - Fe₂O₄ - Cubic - Face-centered - Fd-3m (227) - I/c PDF 5.2 - S-Q 19.4 %
 - 04-007-9753 (*) - Iron - Fe - Cubic - Body-centered - Im-3m (229) - I/c PDF 11.6 - S-Q 12.6 %
 - ▼ 01-081-0462 (I) - Goethite, syn - FeO(OH) - Orthorhombic - Primitive - Pbnm (62) - I/c PDF 3. - S-Q 2.6 %



- G1 6 pin
- 04-007-9753 (*) - Iron - Fe - Cubic - Body-centered - Im-3m (229) - I/c PDF 11.6 - S-Q 55.7 %
 - 01-083-0112 (*) - Magnetite-C, syn - Fe₂1.333O₃2 - Cubic - Primitive - P4332 (212) - I/c PDF 4.5 - S-Q 17.4 %
 - 00-044-1415 (*) - Lepidocrocite, syn - Fe+3O(OH) - Orthorhombic - Base-centered - Bbmm (63) - I/c PDF 1.2 - S-Q 22
 - ▼ 01-081-0462 (I) - Goethite, syn - FeO(OH) - Orthorhombic - Primitive - Pbnm (62) - I/c PDF 3. - S-Q 4.3 %



- G2-1 disc
- 04-007-9753 (*) - Iron - Fe - Cubic - Body-centered - Im-3m (229) - I/c PDF 11.6 - S-Q 92.7 %
 - 04-010-4300 (I) - lepidocrocite - FeO(OH) - Orthorhombic - Base-centered - Cmcmm (63) - I/c PDF 3.6 - S-Q 7.3 %



- G2-1 pin
- 04-007-9753 (*) - Iron - Fe - Cubic - Body-centered - Im-3m (229) - I/c PDF 11.6 - S-Q 94.4 %
 - 04-008-8203 (*) - Manganese Carbide - Mn₂3C₆ - Cubic - Face-centered - Fm-3m (225) - I/c PDF 2.7 - S-Q 5.6 %



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Treatment	Conditions	Generated	Generated from ref.
Sprayed salt solution (1%) for half an hour	40°C, 2 hours with high humidity, 6 hours with low humidity. 5 cycles	$\alpha\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ $\gamma\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ Fe_3O_4 $\gamma\text{-Fe}_2\text{O}_3$	$\alpha\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ $\beta\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ $\gamma\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ Fe_3O_4
No salt solution		$\gamma\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$	$\alpha\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ $\gamma\text{-Fe}_2\text{O}_3\cdot\text{H}_2\text{O}$ Fe_3O_4

環境	条件	発生さび		
純水噴霧 (H ₂ O)	短時間 4.5hr (short)	-	-	γFeOOH
	長時間 20d (long)	αFeOOH Fe_3O_4	-	γFeOOH
	長時間噴霧後加熱 (heat)	- Fe_3O_4	- $\alpha\text{Fe}_2\text{O}_3$	-
3%NaCl 水溶液噴霧 (NaCl)	短時間 4.5hr (short)	-	βFeOOH	γFeOOH
	長時間 20d (long)	αFeOOH	βFeOOH	γFeOOH am. FeOOH
	長時間噴霧後加熱 (heat)	- Fe_3O_4	- $\alpha\text{Fe}_2\text{O}_3$	-

---K. Ohno, Y. Ogawa, RTRI Rep., No. A-83-70, 1983

Pin-on-disc testing:



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Contact pressure: 900MPa;

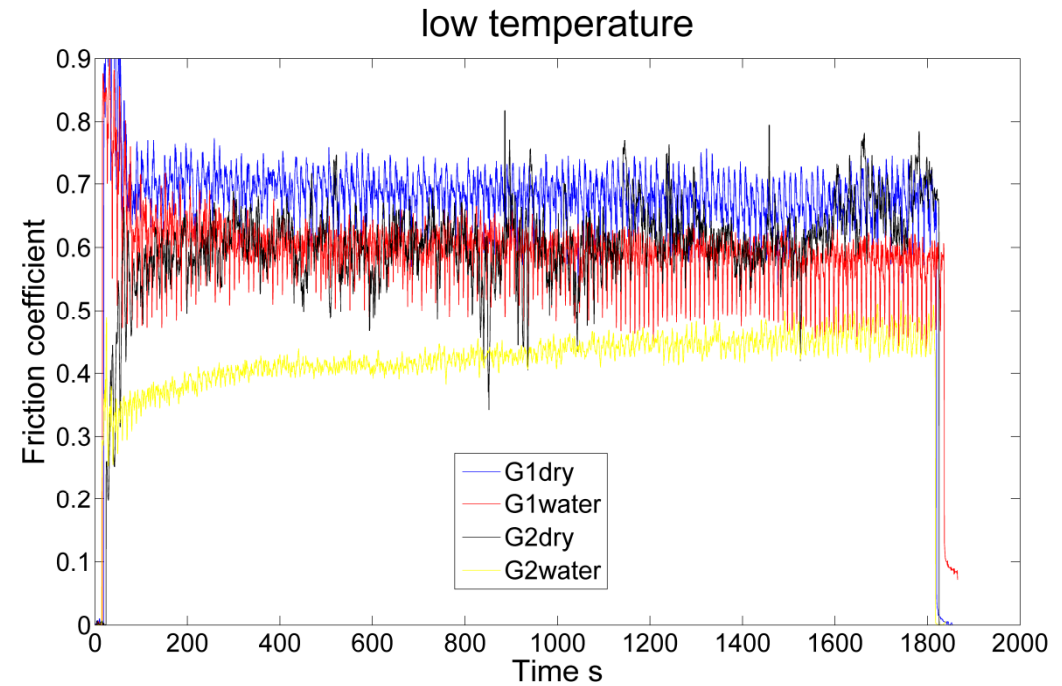
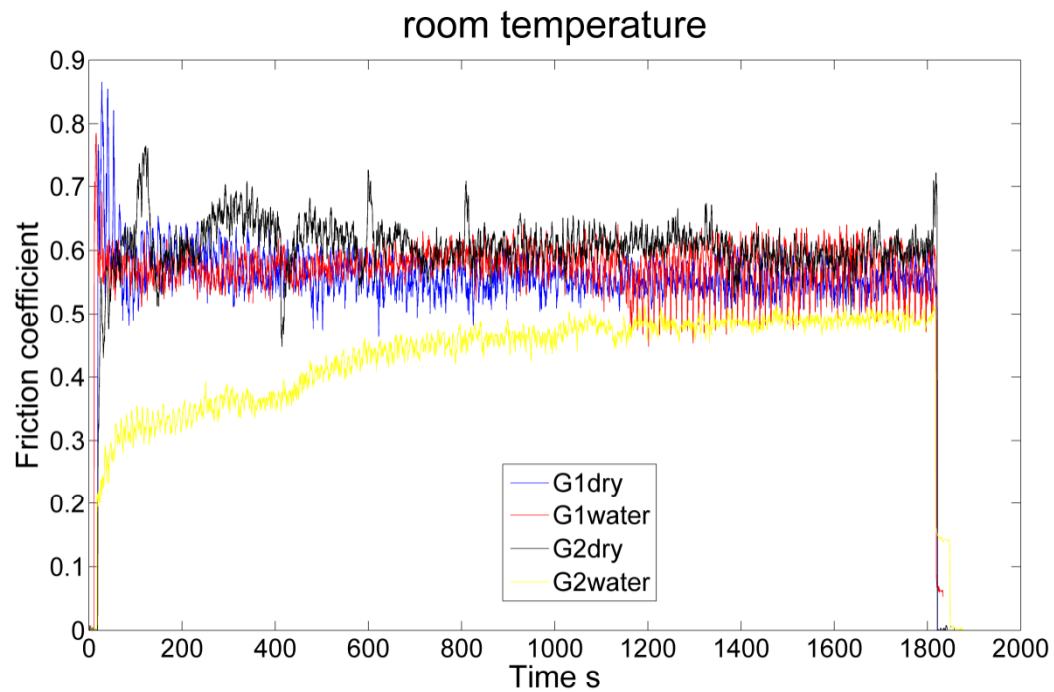
Sliding speed: 0.01m/s

Pin-on-disc testing with two temperature 20°C (+) and 10°C (-) with
RH 40%±5%; with water (w) and dry (d)

Test duration: 30min

Sliding distance: 18m

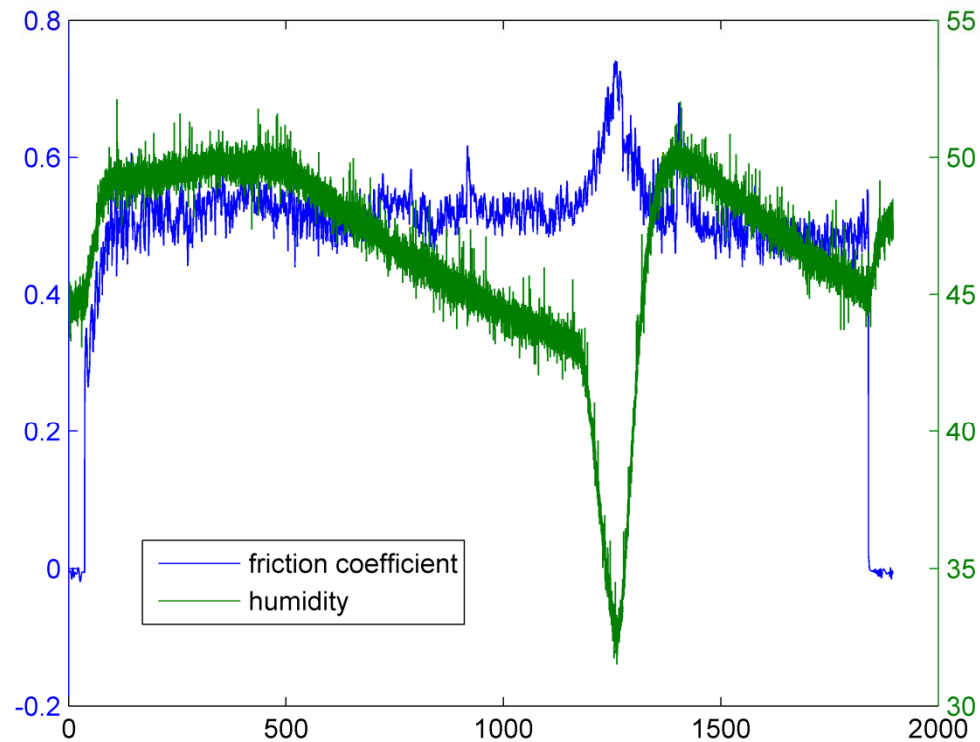
Friction results:



In one of the tests, interesting results are observed!



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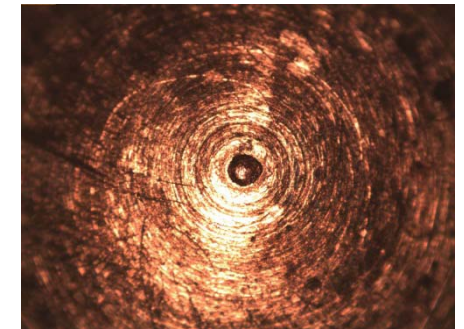
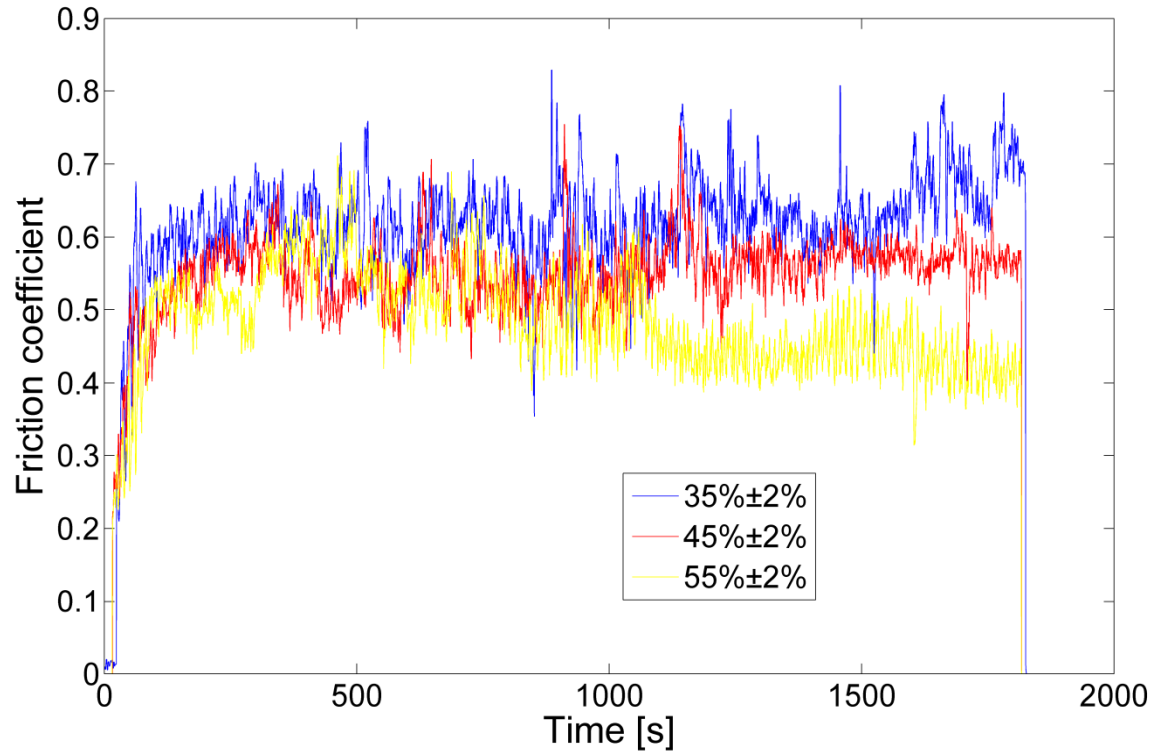


G2 exposure (thin oxide, 10°C):

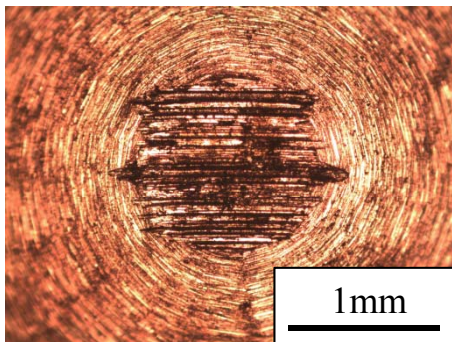
More tests are done with regard to different levels of humidity:
On the thin oxide specimens at 10°C



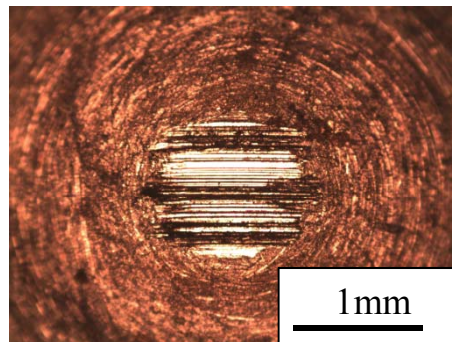
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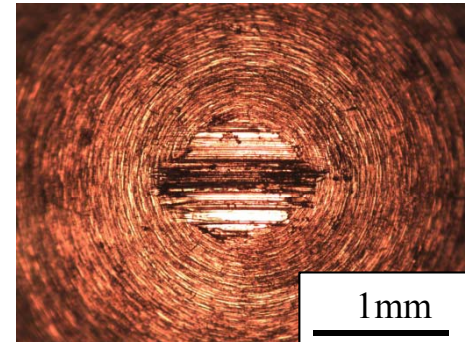
Water lubricated



35%



45%



55%

Specimens	Height [mm]	Width [mm]
35%	1.2	1.5
45%	0.97	1.1
55%	0.79	1.1

[Ref] O. Hayashi et. al, Influence of atmospheric conditions upon adhesion between rails and running wheels, Nippon Kikai Gakkai Ronbunshu, C Hen/ Transactions of Japan Society of Mechanical Engineers, Part C 63 (606), pp. 566-571



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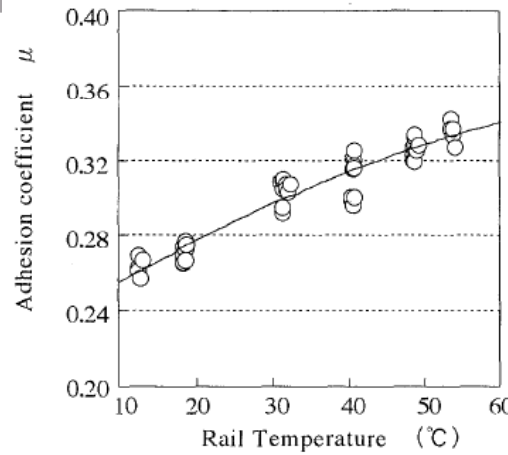


Fig.3 Reference Rail Temperature with μ (Atmosphere Temp. 14°C, Humidity 70%)

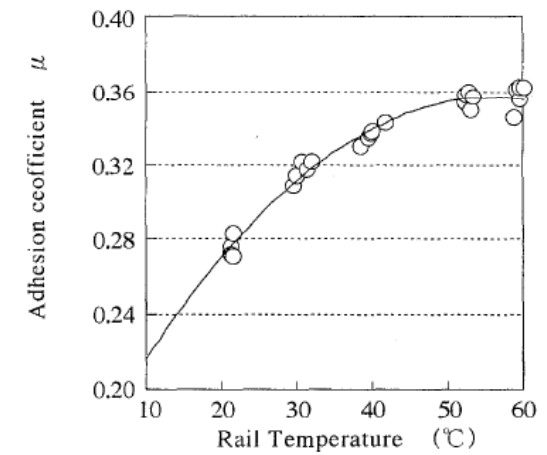


Fig.4 Reference Rail Temperature with μ (Atmosphere Temp. 20°C, Humidity 55%)

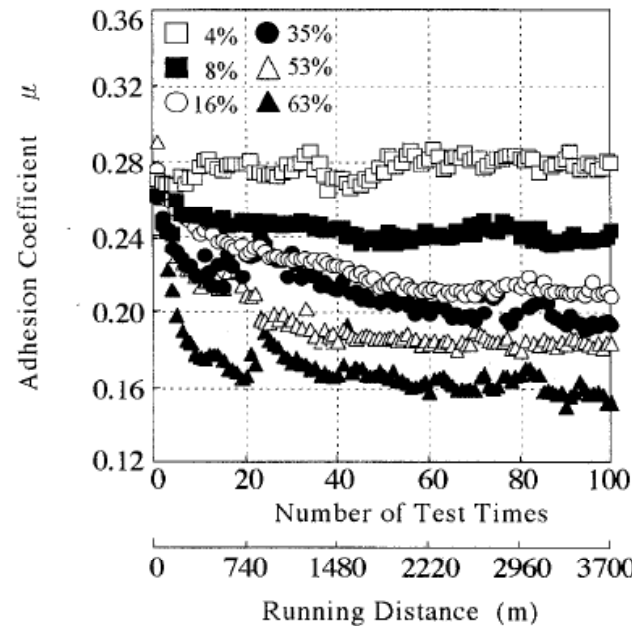


Fig.5 Behavior of μ changed with the Running Distance of Test Truck under Various Humidity of the Atmosphere around the Running Surface of Rail (Atmosphere Temp. 15°C)

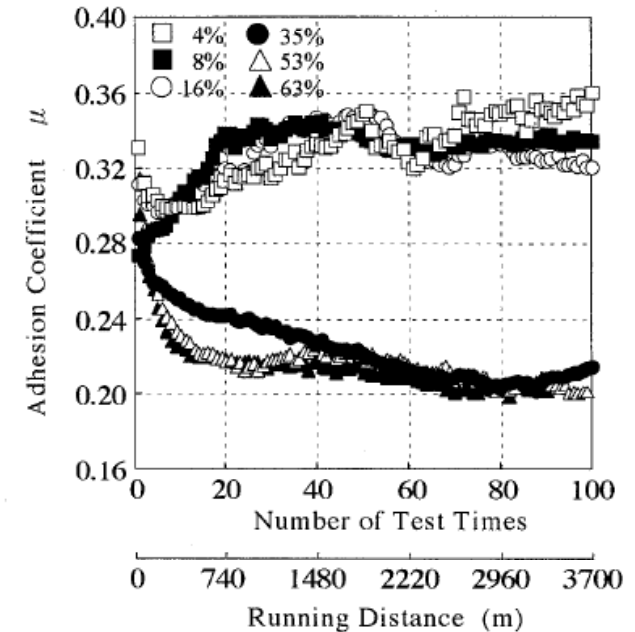


Fig.6 Behavior of μ changed with the Running Distance of Test Truck under Various Humidity of the Atmosphere around the Running Surface of Rail (Atmosphere Temp. 28°C)

Discussion

1 for thick oxide specimens, μ is almost independent of lubricants (dry or water), especially at room temperature. The running-in process is also independent of the environmental conditions. These phenomena are probably due to the extremely rough surface topography.

2 for thin oxide specimens, the running-in process lasts only for a very short while (< 2 min, sliding distance around 1 meter) under dry condition. However, under wet condition, the running-in process lasts long.

3 thin oxide specimens are very sensitive to relative humidity.



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Future work:

1 more tests will be done for thin oxide specimens at various humidity levels; temperature levels.

2 surface analysis



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Thank you for your attention!

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